Exam 2 Sci 265 Fall 2015

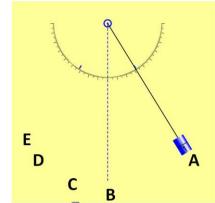
| Name: | Group | : |
|-------|-------|---|
| | | |

| 1. | When Nicole runs | a marathon fron | n the start | of the race t | o the end | when she' | s hugging | her opponent, | , energy | begins as |
|----|------------------|-----------------|-------------|---------------|-----------|-----------|-----------|---------------|----------|-----------|
|----|------------------|-----------------|-------------|---------------|-----------|-----------|-----------|---------------|----------|-----------|

- _____ and ends up as ______.
- A. Chemical, Thermal
- B. Chemical, Kinetic
- C. Kinetic, Thermal
- D. Kinetic, Gravitational Potential
- E. Gravitational Potential, Kinetic

A pendulum is pulled to the side. After it is released, it swings to the other side and back. There is no friction. The lowest point that the Bob can reach is considered the zero potential energy level.

- 2. The highest point that it will reach on the left side is
 - A. It will not swing
 - B.Postion B
 - C. Position C
 - D. Position D
 - E. Position E
- 3. At position B, what type of energy does the pendulum have?
 - A. Kinetic
 - **B.Potential**
 - C. A combination of both Kinetic and Potential
 - D. Thermal
- 4. At position D, what type of energy does the pendulum have?
 - A. Kinetic
 - **B.Potential**
 - C. A combination of both Kinetic and Potential
 - D. Thermal
- 5. If an object carries a net electrical charge, it will
 - A. repel all other objects
 - B. repel all plastic objects
 - C. repel objects with the opposite charge
 - D. repel objects with a like charge
- 6. If an object carries a net electrical charge, it will attract bits of paper
 - A. only if it has a positive charge
 - B. only if it has a negative charge
 - C. if it has either a positive or a negative charge
- 7. An item with a positive charge
 - A. gained protons
 - B. lost electrons
 - C. lost protons
 - D. gained electrons



| 8. | A 11 | magne | |
|----|------|-------|-----|
| × | Δ11 | magni | 210 |
| | | | |

- A. have a net electrical charge
- B. are attracted to all metals
- C. both A and B
- D. neither A or B

9. Energy contains mass

- A. True
- B. False
- 10. A person has a magnet and they cut the North end off and throw it away. Now they are left with a
 - A. South pole
 - B. North pole
 - C. North and a South pole
 - D. It is no longer a magnet

11. In lab, magnets strongly attracted

- A. A nail, paperclip and aluminum rod
- B. A nail and paperclip
- C. plastic pipe, clear plastic and glass test tube
- D. A and C
- E. B and C

12. Power is

- A. energy output per unit time
- B. a form of Energy
- C. a force
- D. All of the above
- E. None of the above
- 13. Consider the 1kg of feathers and 1 kg of gold, the
 - A. 1 kg of gold has a smaller weight
 - B. 1 kg of feathers has a smaller weight
 - C. 1 kg of feathers and 1 kg of gold have equal weight
 - D. Depends on the temperature and pressure of the room.
- 14. What does the density of an object have to be to float in water?
 - A. less dense than water
 - B. more dense than water
 - C. Both of the above
 - D. Density alone is not enough information

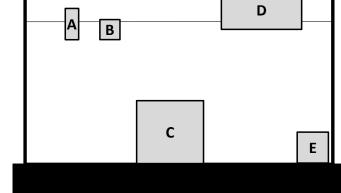
| Low Density | | Water | | | | Hi | gh Density |
|-------------|-----|-------|-----|---|-----|----|------------|
| 0 | 0.5 | 1 | 1.5 | 2 | 2.5 | 3 | 3.5 |

Density (kg/L)

- 15. At the point labeled "1.5" on the Density line above is it most appropriate to write:
 - A. Sinks quickly
 - B. Barely sinks
 - C. Barely floats
 - D. Floats well
 - E. Not enough information

You have 5 blocks of different shapes, different sizes, and different materials. You put them in water and see that some float and some sink (see picture).

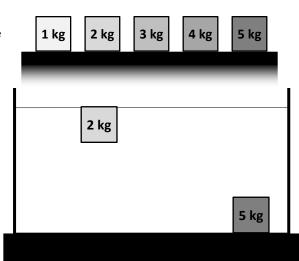
- 16. Which of the blocks has the smallest density?
 - A. Block A
 - B. Block B
 - C. Block D
 - D. Blocks A, B and D
 - E. Blocks C and E
- 17. Which of the blocks has the largest density?
 - A. Block A
 - B. Block B
 - C. Block D
 - D. Could be either blocks A, B or D
 - E. could be either Blocks C or E



- 18. If you compare Blocks C and E, which of the blocks has the largest volume?
 - A. Block C
 - B. Block E
 - C. Blocks C and E have the same volume
 - D. Not enough information

You have 5 blocks that are the same size, but different masses. The lightest one is 1kg, the heaviest one is 5kg. The picture shows how the 2kg and 5kg blocks float and sink in water.

- 19. If you place the 3 Kg block in the water, it will
 - A. float with part of the block above the water
 - B. float with just the top of the block at the water
 - C. float at any depth it is placed
 - D. Sink
 - E. Not enough information



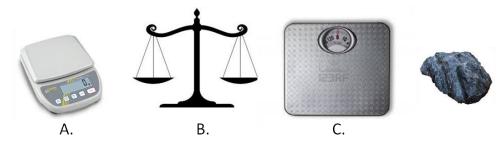
The diagram to the right shows two identical cylinders filled to the same level with water. Also shown are two marbles, one glass and one steel. The two marbles are the **same size** but the steel one is much heavier than the glass one.

GLASS MARRIE

CYLINDER 1

STEEL MARBLE

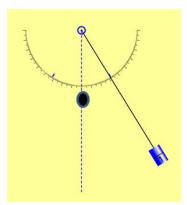
- 20. When the glass marble is put in Cylinder 1 it sinks to the bottom and the water level rises to the 6th mark. If you put the steel marble into Cylinder 2, what will happen? The steel marble will
 - A. float
 - B. sink and the water will rise to the 6th mark.
 - C. sink and the water will rise higher than the 6th mark.
 - D. sink and the water will rise but lower than the 6th mark.
- 21. A person stands on a bathroom type scale on earth and then travels to the moon with the scale. When they stand on the scale on the moon, the scale will read less because
 - A. A person weighs less on the moon
 - B. A person has less mass on the Moon
 - C. A and B
 - D. The scale will not read less
 - E. None of the above



- 22. Which of the devices above will measure this rock the same on the moon compared to on Earth?
 - Δ Δ
 - B. B
 - C. C
 - D. A and C
 - E. None of the above
- 23. Magnesium's symbol is?
 - A. M
 - B. Ma
 - C. Mg
 - D. Mn
- 24. The symbol for Potassium is
 - A. P
 - B. Ph
 - C. Po
 - D. K
- 25. He stands for ...?
 - A. Helium
 - B. Hydrogen
 - C. Hendrox
 - D. Iron

| 26. | Wh | at do the groups/families in the periodic table have in common? |
|-----|------|---|
| | A. | The number of total electrons |
| | В. | Similar masses |
| | C. | Similar characteristics |
| | D. | The number of neutrons |
| | | |
| 27. | The | ability to pound a metal into a flat sheet without breaking it is |
| | A. | Conductivity |
| | В. | Luster |
| | C. | Ductile |
| | D. | Malleability |
| | | |
| 28. | Wh | at tool is used to find mass in the laboratory? |
| | A. | Scale |
| | В. | Meter stick |
| | C. | Graduated cylinder |
| | D. | Balance |
| | | |
| 29. | Wh | at is the base unit for measuring distance in the SI system? |
| | A. | Gram |
| | В. | Liter |
| | C. | Meter |
| | D. | Newton |
| | | |
| 30. | Hov | v many kilograms in 8000 grams? |
| | A. | 0.0008 kg |
| | В. | 8 kg |
| | C. | 80 kg |
| | D. | 8000 kg |
| | | |
| 31. | Wh | ich metric unit would be most convenient for measuring the length of a ladybug? |
| | A. | Millimeter |
| | В. | Centimeter |
| | C. | Meter |
| | D. | Gram |
| | | |
| 32. | Of t | he choices below, which represents the largest SI prefex? |
| | A. | centi |
| | В. | milli |
| | C. | Mega |
| | D. | Tera |
| | | |
| 33. | | symbol Ag stands for which element? |
| | Α. | Copper |
| | В. | Silver |
| | C. | Gold |
| | D. | Mercury |
| | | |
| | | |

34. A pendulum made from a string and a mass starts from rest as shown. There is a barrier placed at the black dot. Draw below *on the same diagram* how the pendulum will look when it has swung as far left as it can go. Draw both the string and the bob. Be very clear about the final height (on the left) compared to the original height (on the right).



35. Cameron decided she wanted to figure out the amount of horsepower she could produce in a short burst so she timed how long it took her to sprint up a flight of stairs. She measured the height difference from the ground floor to the next level to be 210 cm. She has a mass of 50 kg and was able to sprint to the top in 1.8 seconds. Useful equations: Potential energy = mass*gravity*height, Power = Energy/time. Useful data: g = 9.8 m/s^2, 746 Watts = 1 hp. Use the factor label method for all conversions.

36. How is the arrangement of items in the supermarket similar to the arrangement of the elements on the periodic table?

